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Chapter 14 Acids And Bases

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an acid that is composed of H, O, and a third element. The H & O are covalently bonded and are called the hydroxyl group. Property of bases. taste bitter, feel slippery, change color of indicators, react with acids to produce a salt and water neutralization rxn, are electrolytes. Arrhenius or traditional base.

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(14.1) The nature of acids and bases
(14.2) Acid strength (14.3) The pH scale
(14.4) Calculating the pH of strong acid
solutions (14.5) Calculating the pH of
weak acid solutions (14.6) Bases (14.7)
Polyprotic acids

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Chapter 14 - Acids and Bases . 14.1 The Nature of Acids and Bases . A. Arrhenius Model 1. Acids produce hydrogen ions in aqueous solutions 2. Bases produce hydroxide ions in aqueous solutions B. Bronsted-Lowry Model 1. Acids are proton donors 2. Bases are proton acceptors 3. H. 3. O + is called the

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hydronium ion C. Conjugate Acid- Base Pairs 1.

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Chapter 14 - Acids and Bases.

Description. Chapter 14 Review. Total
Cards. 36. Subject. Chemistry. Level.
11th Grade. ... General Properties of

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Acids: Definition. 1) sour taste, corrosive
2) characteristically changes acid-base
indicator 3) react with active metals to
form hydrogen gas 4) react with bases
to form salt and water 5) conduct ...

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ANS: e) ClO_4^- , F^- , OAc^- , CN^- PAGE: 14.2

11. The conjugate base of a weak acid is

a) a strong base b) a weak base c) a

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strong acid d) a weak acid e) none of these ANS: b) a weak base PAGE: 14.2
12. Which of the following is the equilibrium constant expression for the dissociation of the weak acid HOCl? a) $K = \frac{[H^+][OCl^-]}{[HOCl]}$

CHAPTER 14 Acids and Bases

Modern Chemistry Chapter 14 Vocab

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Acids and Bases. binary acid. oxyacid.
Arrhenius acid. Arrhenius base. an acid
that contains only two different
elements: hydrogen an.... an acid that is
a compound of hydrogen, oxygen, and a
third el.... a chemical compound that
increases the concentration of hydrog....

chemistry chapter 14 acids bases

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CHAPTER 14 REVIEW Acids and Bases

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SECTION 1 SHORT ANSWER Answer the following questions in the space provided.

1. Name the following compounds as acids:

- a. H_2SO_4 sulfuric acid
- b. H_2SO_3 sulfurous acid
- c. H_2S hydrosulfuric acid
- d. HClO_4 perchloric acid
- e. HCN hydrocyanic acid

2. H^+ hydrogen cyanide

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14 Acids and Bases - Password

acid, and water acts as the base. This is the acid ionization reaction. $HA(aq) + H_2O(l) \rightleftharpoons A^-(aq) + H_3O^+(aq)$ 6) When a base is added to water, there is a Bronsted-Lowry acid base reaction where the base behave as a base and water acts as the acid. This is the base ionization reaction. Be able to identify or

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provide the conjugate acids and ...

Chapter 14: Acids and Bases

CHAPTER 14 ACIDS AND BASES 501 24.

A Lewis acid must have an empty orbital to accept an electron pair, and a Lewis base must have an unshared pair of electrons. 25. $\text{NH}_3 + \text{NH}_3 \rightleftharpoons \text{NH}_2^- + \text{NH}_4^+$
Acid Base Conjugate Conjugate

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Base Acid

CHAPTER 14 ACIDS AND BASES - HCC Learning Web

Major topics: Arrhenius vs. Bronsted-Lowry definition of acids and bases, conjugate acid/base, acid dissociation constant (K_a), & strong vs weak acids

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Chapter 14 (Acids and Bases) - Part 1 - YouTube

Chapter Fourteen: Acids and Bases

Chapter interactives: Acids/bases and
indicators Chemistry battleship

Chemistry jokes pH simulation Chapter

Homework: Section 1: Chapter review 1

thru 7. Section 2: Chapter review 12, 13.

Section 3: Chapter review 19 thru 25.

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Chapter Fourteen [Acids and Bases]

Chapter 14: Acids and Bases

1. Acids and Bases. 1. Which of the following is the equilibrium constant expression for the dissociation of the weak acid HOCl? e) none of these.

2. In which of the following reactions does the H_2PO_4 ion act as an acid? d) The ion cannot act as

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an acid.

Chapter 14: Acids and Bases 1 - Docest

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Chapter 14 - Acids and Bases

Chapter-14-Acids-and-Bases - CHAPTER
14 ACIDS AND BASES In this chapter you
will learn The definitions and identity of
substances that are called acids

Chapter-14-Acids-and-Bases - CHAPTER
14 ACIDS AND BASES In...

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Chapter-14-Acids-and-Bases - CHAPTER 14 ACIDS AND BASES In ...

Acids and Bases: Chapter 14 & 15 . HW:
•Read Ch 14: •Fill in as much of the acid
base table as you can, as you read . Acid
base conductivity and reactivity
Conducvity, Reac'vity,,
Hydrochloric*acid* high high
Acidic*acid* ** low* medium

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Distilled*water* none* none*

Ammoniumhydroxide med* none*

Sodiumhydroxide* high none* *

Acids and Bases: Chapter 14 & 15

Friday, January 19 - Day A Agenda: 1)
Test - Chapter 12 and 13 HW: 1) Watch
Videos and take notes on intro to acids
and bases , and pH scale 2) Read 14.1-

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14.4

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