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law of conservation of mass (8.1) The principle that the total amount of matter is constant during any physical or chemical change (matter is neither created nor destroyed during these types of changes).

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Modern chemistry homework 8-1 answers - What is The ...

Modern Chemistry Chapter 8. chemical equation. skeleton equation. balanced equation. precipitate. a concise representation of a chemical reaction using chemical.... a chemical equation that does not indicate the relative amount.... an equation in which each side has the same number of atoms....

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Modern Chemistry Chapter 8. chemical equation (8.1) precipitate (8.1) coefficient (8.1) word equation (8.1) a way to describe a chemical reaction using chemical formulas.... a solid that forms from a solution during a chemical reaction. A number in front of a chemical formula in an equation that in....

notes chapter 8 modern chemistry Flashcards and Study Sets ...

CHAPTER 8 REVIEW Class 5Ec7IB-n Chemical Equations and Reactions SHORT ANSWER Answer the following questions in the space provided. 1. Match the symbol on the left with its appropriate description on the right. (aq) (a) (b) (c) (d) (f) A precipitate forms. A gas forms. A reversible reaction occurs. Heat is applied to the reactants.

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Write the chemical equation that relates to each of the ...

CHAPTER 8 REVIEW Chemical Equations and Reactions MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. b A balanced chemical equation represents all the following except (a) experimentally established facts. (b) the mechanism by which reactants combine to form products.

8 Chemical Equations and Reactions

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Write the balanced equation for each of the following: a ...

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How many atoms of each type are represented in ... - StudySoup

Modern Chemistry • CHAPTER 8 HOMEWORK 8-3 (pp 248–250) VOCABULARY Complete each sentence 1 A chemical equation is like an algebraic equation because they both ____ 2 The coefficients of a chemical reaction indicate ____ 3 A chemical

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