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For questions 1 - 4 an amount of solute is given, and a ... For

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questions 11 - 12,
show your work and
circle your final
answer. 11. If 115 g
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100 g of water at 35
°C, how ...

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solubility. _____ 3.

According to the graph,
the solubility of any
substance changes as
_____ changes. 4.

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Solubility Graph
Worksheet 1. Why do the temperatures on the graph only go from 0°C to 100°C ? Water changes to a gas at

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100° C and will no longer dissolve solute as a gas. 2. Which substance is most soluble at 60° C? potassium iodide 3. Which two substances have the same solubility at 80° C ? _ sodium chloride, potassium chlorate

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According to the graph,
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The curve of a solubility graph represents the saturation point for the solute at increasing temperatures. Above the line, any dissolved solute is a supersaturated

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solution Below the line,
any dissolved solute is
an unsaturated
solution. SAMPLE

QUESTION: Which term
describes a solution
containing 30 g KNO_3
dissolved at 40°C ?

ANSWER: unsaturated

Reading solubility graphs

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Interpreting Solubility
Curves How to read a
solubility curve?

Example: Refer to
graph to answer the
following questions: 1.

What mass of
Ammonium Chloride
will dissolve at 50°C in
100 g of water? 2.

What is less soluble in
100 g of water at 10°C
sodium nitrate or
sodium chloride? 3.

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